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chemical reactions, but

they are cumbersome.

It's easier to use the

formulas for the

reactants and products

to write chemical

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equation is a

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representation of a  
chemical reaction; the  
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The formulas of the

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reactants on. Moreover you must cite properties that change as a consequence of the reaction.

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Hindenburg was heading for its landing site in Lakehurst, New Jersey, after an uneventful trans-Atlantic crossing. Suddenly, to the horror of observers on

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SECTION 11.1  
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for the reaction  
between hydrogen and  
oxygen that 111  
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Answers if a  $\Delta$  is used  
to increase the rate of  
a chemical reaction, its  
formula is written  
above the arrow ST

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SECTION 11.1  
DESCRIBING CHEMICAL  
REACTIONS (pages

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## PDF Section 11.1

321-329) This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.

### **SECTION 11.1** **DESCRIBING** **CHEMICAL**

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**REACTIONS (pages  
321-329)**

Section 11.1 -  
Describing Chemical  
Reactions In a chemical  
reaction, the reactants  
are written on the left  
and the products on  
the right. The arrow  
that separates them is  
called yield. Reactants  
□ Products

**Chapter 11:  
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**...Chemical**

Section 111 Describing  
Chemical Reactions

Pages 321 329 ... In a chemical reaction, one or more reactants change into one or more products. Figure 11.1a shows the ingredients for making leavened bread— flour, salt, yeast, and water. Chemical reactions take place when the ingredients are mixed together and heated in

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the oven.

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they are cumbersome.  
It's easier to use the  
formulas for the  
reactants and products  
to write chemical

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equations. A chemical equation is a representation of a chemical reaction; the formulas of the reactants (on

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Word equations  
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are temporary and self-  
limiting and do not  
need medication.  
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cite properties that

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change as a  
consequence of the  
reaction.

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Worksheet Answers**

322 Chapter 11 Figure  
11.2 Three common  
chemical reactions are  
shown. When methane  
gas burns, it combines  
with oxygen to form  
carbon dioxide and  
water. Iron turns to red-  
brown rust (iron(III)  
oxide) in the presence

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of oxygen in the air.  
Hydrogen peroxide  
decomposes to water  
and oxygen when used  
as an antiseptic.

### **11.1 Describing Chemical Reactions 11**

Chapter 11 Chemical  
Reactions 113 SECTION  
11.1 DESCRIBING  
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REACTIONS... A  
chemical reaction  
occurs when one or  
more \_\_\_\_\_ change into

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one 8 CHEMICAL REACTIONS - Lake Travis ISD SECTION 8.1 DESCRIBING CHEMICAL CHANGE (pages 203211) This section explains how to write equations describing chemical reactions...

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